

VISVODAYA GOVERNMENT DEGREE COLLEGE

VENKATAGIRI, TIRUPATI DISTRICT, ANDHRAPRADESH.

Criteria – I – Curricular Aspects

1.1 Curricular Planning and Implementation

1.1.1 The Institution ensures effective curriculum planning and delivery through a well-planned and documented process

Departmental Activity Register

			Proforma	for Departmenta	LActivity Dogiet	or			
Nan	no of the l	0	Froitima	101 Departmenta	Activity Regist	C1			
		Department: physics					Academic		2022
S.NO	Date	Conducted through DRC/JKC/NCC/NSS/Department		Title of the Activity	Name of the Lecturer(s) involved	Details of the Resource person	No. of students participated	Signature of the Dept., Incharge	Remark
1.	17.11.21	DEPARTMENT	(CLASS TEST)	CLASS TEST	N-Nagokaj	-	IseH()	(1)	
2	. 24.11-21	DEPARTHENT	CO-CURRICULAR	STUDENT	N. Nagarai	-	V 584 (2)	(1) <u>13</u> .	
3	06-12-21	DEPART HENT	CURRICULAR	CLASS TEST	N. Naguras	-	2 ser ()	(n) 1/2	
4	18-12-21	DEPARTMENT	CURRICULAR	CLASS 745T	N. Nagaraj	-	V SON()	13	
5.	29.12.21	DEPARTHENT	CURRICULAR	CLASS TEST	N. Nagaraj	-	I SEH (11)	(1) A	F
6.	06-01-22	DEPARTMENT	CURRICULAR	CLASS TEST	N. Nagaraj	-	inser()	1/2	
7.	07.01-22	DEPARTHENT	CURRICULAR	CLASS TEST	N. Nagarai	-	QSEM()	(1/2	
			SANKRE	INT HOLIDAPS	FROM 10.0	1. 22 to	15. 01.22		
8-	20-01-22	DEPARTMENT	CURRICULAR	CLASS TEST	N. Nagaraj	-	SEM()	15	
9.	03-02.22	DEPARTHENT	CURRICULAR	CLASS TEST	N. Nagaraj		& SEM()	(1/2	
10-	04.02.22	DEPARTMENT	CURRICULAR	CLASS TEST	N. Nagaraj	-	E SEH ()	0/2	
13.	07-02-22	DEPORTMENT	CURRICULAR	CLASS TEST	N. Nagaroj	-	9 SETP1	(n /s	
12.	07-02-22	DEPARTMENT	CURRICUCAR	CLASS TEST	N. Nagarej	-	in sen	() z	
1	64			CA	,				

Annual Curricular Plans

VISVODAYA GOVT. DEGREE COLLEGE

	1	1	c (B.Z.C) V Semes To Year: 20							ogy (Paper					_
		s	1-1	Additional Input/	Cu		ır Acti			Extra/Co-				_	8
Month	Week	Hours	Syllabus Topic	Value Addition Provided/ taught	Activity Conducted	Hours	Whether Conducted	Whether Conducted	If not alternate	Activity Conducted	Hours	Whether	Whether Conducted	If not alternate	Remarks
January 2082	ist.	T P	Fermination - types - Submorged Continuous atimed link Arolist Fixed Bed and Fluidized	l av	Assignment 100-5		yes		- 11	Field Thip		Yes			
Эапнаву 2022	and	7 P 3+2	chromate graphy and yophiliantion	1 1	Internal Exam-2	t	Yes	Panel		hab cleaning	l Hour	Yes			
Emmoy 2022	399	T P 3+2	Agriculture Hehemics - Monaultur Peng Producty, Drug finger printing		Prefinal Exam Hedel Ponchal Exam	3	Yes	mg 2		para ell					a
Janussy 8022	7 ^{TL}	7 P 312	Revision of the sylledou Reful 288 Commentement of Theory Exams		(-= 1	- Ma	la U	Pine					6		
Tanuary 2022	514			6) 44	1		Hartl	2023	1	15 d x 14 30 o 7 14					

VISVODAYA GOVT. DEGREE COLLEGE Annual Curricular Plan (Lecturer Wise): 2020-2021

		9	3	Additional	Cur	ricula	r Activ	ity		Co-C	urricul	arAct	tivity		S
Month	Week	Hours Available	Syllabus Topic	Input/ Value Addition Provided/ taught	Activity Conducted	Hours Allotted	Whether Conducted	Whether Conducted	If not alternate	Activity Conducted	Hours A@otted	Whether Conducted	Whether Conducted	If not alternate	Remarks
O CTOBER	1	-		Fales		=84 -4 \	=30 	first fig.	200	onu si		[m]	1		
	2	उपि २०९	Orbits - veelet Jon model-	Structurer of an atom.	Teaching	02	yes	4.0	100	1.51.00	e L	100			
	3	4 (TH) 2 (PR)	Stern-Gerlach experiment -	1	Teaching	03	485	1. 1	10.59	Class Test	01	121	405	- 13	
	4	4 Evi 2 [PO	Raman effect - Quantum likely of Raman effect - Exp., archangement - Applications of Raman effect.	About Matter.	Teaching	04	yes			. Wage To	AT THE	Ð-1		qr 100	
	5	4(tr) 2(pr)	Try po the bis - Properties of matter		Teaching	05	705	4.1	6 3 (3) (3) (3)	STUDENT	ol	400	5		

Signature of the Lecturer

Signature of the Department in Charge

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Usage of Teaching – Learning Process (TLP) Mobile App for reporting of curriculum transactions.

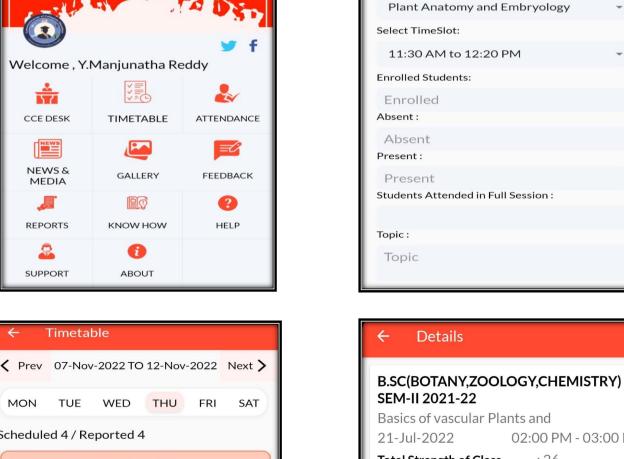
Attendance Entry

II BZC_SEM_III_2022-2023

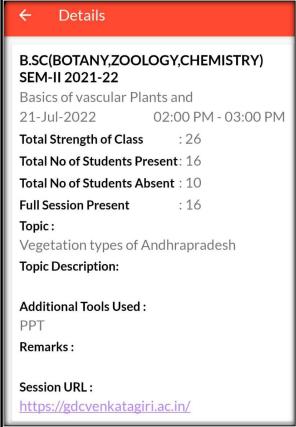
11-Aug-2022 Select Course:

Select Subject:









Teaching diaries

VISVODAYA GOVT. DEGREE COLLEGE, VENKATAGIRI.

Honth-January 2022 Teaching Diary: 2020-2021 Name of the Department /Subject: 20060 Gy Name of the Lecturer : No. of Students Students Methodo Period/ Theory/ Remarks Teaching Activity Conducted Class Medium Topic Covered Day logy Adopted Time Attended Al-aurelion EH STH 19 19/1/22 Theory wed Reproductive Tochnologice BluckBra class preparation co-12.0 Health and Hygiene (MPCs 2 B2C) Theosy EMETH Retairstions BlockBood hedine 1Sc-R.00-3-00 EHSTH Bretigal 126 Trooms genic animals Bleck Board Rtarestion 20 Thu 20/1/22 EMPTH Theory complementing & Supplementing gener Block Board Putawalions EMETH Theory hochizo Embryo bransfex \$26 00-50 EN & TH Theory BlockBoard Ritguestions Phi 21 1 22 Dominum) epistisse & necessive caistasis Locline aı Adamstion 11-00-1200 EMETH Embryo transfer and embryo clowing ochire 4300 Lab work solution Proporation (Pamaldohyde) 596

D Sujal For Signature of the Lecturer Signature of the Department In Charge

Signature of the Principal

VISVODAYA GOVT. DEGREE COLLEGE, VENKATAGIRI.

Teaching Diary : 2021-2022

Name of the Department /Subject: PHYSIC S

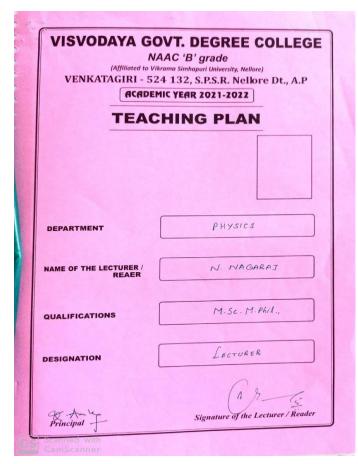
N. Nagaraj Name of the Lecturer: Methodo Period/ Theory/ Practical Class Medium **Topic Covered** logy Adopted Teaching Adis used Time 9 preparation for class work Adiabatic demagnetzation y Liquificata 8 Bsc Compali 15N6 Sorter with Students on cram post 01.02.201.02 Boaks ш́ Вs 10 preparation of MP Chiteria for NADO V 411 pheparation for class work. Teaching Gusty Board 8 Revision on Hatter waves 05 Theory ių Prepulation for class work. ONE 0 02.02 Teaching Questing Compality Liquification of helium 20 I Bs. preparation of evidences for 2st Crifteria 2 4,09 preparation for class work Teachin il As Exam on Heinsner effect Theory 1 phinciple of Refrigeration. Effect of Cfcs of Oxone layer, approars of lin by NAME Record worse CIULS Compute iii 3. THU 03-02 P Record worm for NOOC 28vi

Signature of the Lecturer

Signature of the Department In Charge

Signature of the Principal

Teaching plans & Synopsis



	Collegiate Education, A.P. aching Plan
ame of the Department/Subject	Physics
ame of the Lecturer	N.Nagaraj
ourse/Group	II B.Sc (MPCs)
iner	SEM-IV (Thermodynamics)
ame of the topic	Kinetic theory of gases
ours required	10
earning objectives	Postulates of kinetic theory of gases, Maxwell's law of molecular speeds, Transport phenomenon.
revious knowledge to be remembered Opic Synopsis:	Average velocity, rms velocity
container. The basic version of the model describe particles and, thus, the nature of kinetic energy transf	the macroscopic properties of gases, such as volume, pressure, and
container. The basic version of the model describe particles and, thus, the nature of kinetic energy transparent and the seminary of the semin	lisions between themselves and with the enclosing wan or une set heideal gas and considers no other interactions between the fers during collisions is strictly thermal. the macroscopic properties of gases, such as volume, pressure, and as viscosity, thermal conductivity and mass diffusivity. The model ian motion. mptions: wan as molecules. This smallness of their size is such that the didded up is negligible compared to the volume of the smallest open alent to stating that the average distance separating the gas particles icial treatment can be applied. Ite among themselves and with the walls of the container. All these templecules are considered to be perfectly spherical in shape and ong molecules are negligible. (That is, they exert no forces on one
container. The basic version of the model describe particles and, thus, the nature of kinetic energy transite particles and, thus, the nature of kinetic energy transite and a search of the related phenomena, such as Brown. The theory for ideal gases makes the following assured to the related plane of the individual gas molecules as half containing all the molecules. This is equivated to their size. These particles have the same mass. The number of molecules is so large that statist. The rapidly moving particles constantly collid collisions are perfectly elastic. This means the elastic in nature. Except during collisions, the interactions amonother.)	lisions between themselves and with the enclosing wan of the set heideal gas and considers no other interactions between the fers during collisions is strictly thermal. the macroscopic properties of gases, such as volume, pressure, and sylicosity, thermal conductivity and mass diffusivity. The model iain motion mptions: way as molecules. This smallness of their size is such that the dided up is negligible compared to the volume of the smallest open alent to stating that the average distance separating the gas particles ical treatment can be applied. It among themselves and with the walls of the container. All these molecules are considered to be perfectly spherical in shape and long molecules are negligible. (That is, they exert no forces on one
container. The basic version of the model describe particles and, thus, the nature of kinetic energy transparent properties and the semigrature, as well as transport properties such a alea accounts for related phenomena, such as Brown The theory for ideal gases makes the following assure. The gas consists of very small particles kno total volume of the individual gas molecules a ball containing all the molecules. This is equivis is large compared to their size. These particles have the same mass. The number of molecules is so large that statist. The rapidly moving particles constantly collid collisions are perfectly elastic. This means the dastic in nature. Except during collisions, the interactions amo another.) Examples/Illustrations	lisions between themselves and with the enclosing wan or the set heideal gas and considers no other interactions between the fers during collisions is strictly thermal. the macroscopic properties of gases, such as volume, pressure, and sylicosity, thermal conductivity and mass diffusivity. The model iain motion mptions: way as molecules. This smallness of their size is such that the dided up is negligible compared to the volume of the smallest open alent to stating that the average distance separating the gas particles ical treatment can be applied. It among themselves and with the walls of the container. All these ae molecules are considered to be perfectly spherical in shape and long molecules are considered to be perfectly spherical in shape and long molecules are negligible. (That is, they exert no forces on one
container. The basic version of the model describe particles and, thus, the nature of kinetic energy transite particles and, thus, the nature of kinetic energy transite and a search of the related phenomena, such as Brown. The theory for ideal gases makes the following assured to the related plane of the individual gas molecules as half containing all the molecules. This is equivated to their size. These particles have the same mass. The number of molecules is so large that statist. The rapidly moving particles constantly collid collisions are perfectly elastic. This means the elastic in nature. Except during collisions, the interactions amonother.)	lisions between themselves and with the enclosing wan or the set heideal gas and considers no other interactions between the fers during collisions is strictly thermal. the macroscopic properties of gases, such as volume, pressure, and sylicosity, thermal conductivity and mass diffusivity. The model iain motion mptions: way as molecules. This smallness of their size is such that the dided up is negligible compared to the volume of the smallest open alent to stating that the average distance separating the gas particles ical treatment can be applied. It among themselves and with the walls of the container. All these ae molecules are considered to be perfectly spherical in shape and long molecules are considered to be perfectly spherical in shape and long molecules are negligible. (That is, they exert no forces on one
container. The basic version of the model describe particles and, thus, the nature of kinetic energy transitions and thus, the nature of kinetic energy transitions are the properties such a aboa accounts for related phenomena, such as Byrom. The theory for ideal gases makes the following assured to the such as the properties of the model of	ilisions between themselves and with the enclosing wans on the set heideal gas, and considers no other interactions between the fers during collisions is strictly thermal. The macroscopic properties of gases, such as volume, pressure, and as viscosity, thermal conductivity and mass diffusivity. The model ian motion. The model is an interaction of the model is a motion. The model is an encion. The model is a motion of the model is a motion of the model is a model of the model
container. The basic version of the model describe particles and, thus, the nature of kinetic energy transparent properties and the second properties such a size accounts for related phenomena, such as Brown. The theory for ideal gases makes the following assure. The gas consists of very small particles knot total volume of the individual gas molecules as ability containing all the molecules. This is equivated in the properties of the properties of the properties. The particles have the same mass. The number of molecules is so large that statist. The rapidly moving particles constantly collidicallisions are perfectly elastic. This means the slatic in nature. Except during collisions, the interactions amounther.) Examples/Illustrations Additional inputs Teaching aids used	lisions between themselves and with the enclosing wan on the set heideal gas, and considers no other interactions between the fers during collisions is strictly thermal. The macroscopic properties of gases, such as volume, pressure, and s viscosity, thermal conductivity and mass diffusivity. The model iata medion. The model is modeled to the macroscopic strictly and mass diffusivity. The model is modeled to the macroscopic strictly and mass diffusivity. The model is modeled to the work of the smallest open alent to stating that the average distance separating the gas particles ical treatment can be applied. The among themselves and with the walls of the container. All these temolecules are considered to be perfectly spherical in shape and ong molecules are negligible. (That is, they exert no forces on one of the modeled of the mode

	VT. DEGREE COLLEGE, VENKATAGIRI aching Plan & Synopsis
Name of the Department / St	abject: Zoology
Name of the Lecturer : Dr-	D. Sujatha
Course / Group: ग्रिंग्ने हि.	Sc(BZC) Somestor I
Paper: Paper & Animal	00
Name of the Topic :	Framentalia, Defferent lype of Fermentalia
Hours required :	3 70008
Learning Objectives :	Submerged, & Bolid elate, Food butch
Previous knowledge to be reminded :	Fermulter
Topic Synopsis	(Continue on the reverse side if needed) > From entitle is the process servething the touchemist activity of organisms during their growth, development, reproduct even senescenso and death. > From entitle to the logy is the live of organisms to produce food, pharmaceuticute and decompetering as one drope sade includied baces
Examoles / Illustrations :	Industrial Feomentation
Additional inputs :	upstream proceeding
Teaching Aids used :	Black Board, digital Board
References cited:	Virvek (EM) Animal Bistechnology lost Book.
Student Activity planned after the Teaching :	Put auestoni
Activity planned outside the Class room, if any :	to know the deflevent lypes of
Any other activity :	To observe The mechanism of Fermentalia
· W. Ang	D. Sujaltan Signature of the Lecturer

	T. DEGREE COLLEGE, VENKATAGIRI						
Name of the Department / Subject: Zoology							
Name of the Lecturer: Dr. D. Sujette							
Course / Group: Frod 8 Sc							
	Bartanology						
Name of the Topic :	Toansgenic animale						
Hours required :	2						
Learning Objectives :	Transgenic fish, transgenic goat						
Previous knowledge to be reminded :	Grade harmone, Decomparemt protein						
Topic Synopsis	(Continue on the reverse side if needed) Toursgeric Great: The Sheep the Great Packagenic animal beas used to produce a Decombinant pretion dang in how milk. Basing on this technic Norica Bishimdag. Isomospored the Sill gene from Spriders wint goat the Ossalling mula goats whose to size Silk producing bemula goats.						
Examoles / Illustrations :	Toursgenic organisms Founsgenoses						
Additional inputs :	settical issues , Advantages of						
Teaching Aids used :	Black Board, digital Board						
References cited :	urvek (EM) Arimal Ridectionology teset Book.						
Student Activity planned after the Teaching :	Put austion						
Activity planned outside the Class room, if any :	to observe homegenic animals						
Any other activity :	to partice disoporme						
K. Kmy	D. Swiether Signature of the Lecturer						

ICT Enabled Teaching







Learning by Doing







Field visits
Botany department field visit to Jayaramaiah nursery



Zoology department field visit to Swarna Milk diary



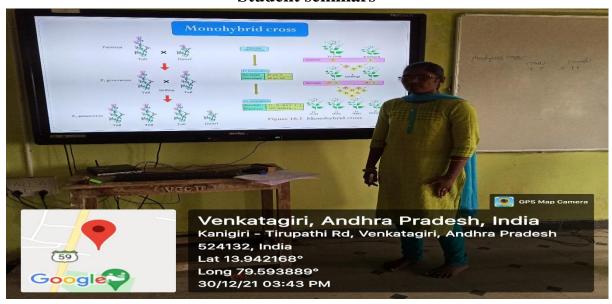
Physics department Educational tour to Regional Science Centre



History department field visit to Mahabalipuram



Student seminars







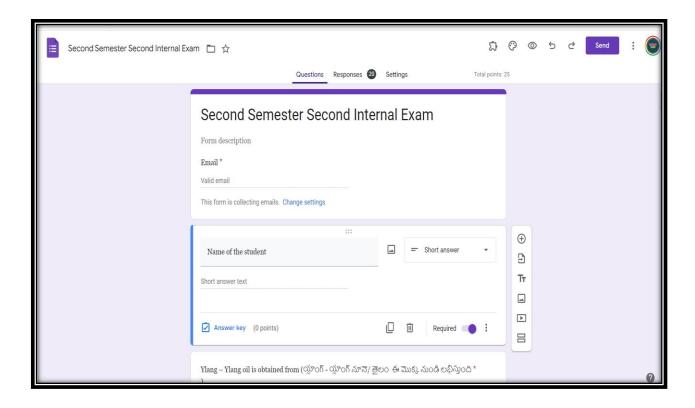
Student Quiz programmes



Group discussions



Online assessment through Google forms





PRINCIPAL)
Visvedaya Govt. Degree College
Venkatagiri - 524132
Tirupati Dt. A.P.